

## Basic Stoichiometry Phet Lab Answers

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Stoichiometry Lab: Data Collection and Processing. IB CHEMISTRY Stoichiometry Lab Data Collection and Processing Item | Mass | Small beaker (100 mL) | 47.0 grams | Large beaker (150 mL) | 82.4 grams | Mass of filter paper | 0.50 grams | Mass of coffee filter | 1.00 gram | 150mL beaker + 20mL water + lead nitrate solution | 96.1 grams | 100mL beaker + 20mL water + sodium carbonate solution | 64.2 grams | Watch glass | 32.2 grams | Precipitate + filter paper + coffee filter | 2.20 grams | ...

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Names: \_\_\_\_ Period: \_\_\_\_ Basic Stoichiometry PhET Lab Let's make some sandwiches! Introduction: When we bake or cook something, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar.

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stoichiometry When the reactants are present in the correct amounts, the reaction will produce products. What happens if there are more or less of some of the reactants present?

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Stoichiometry Basic Introduction, Mole to Mole, Grams to Grams, Mole Ratio Practice Problems - Duration: 25:16. The Organic Chemistry Tutor 614,388 views

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Learn the basics of the Phet lab and worksheet. Learn the basics of the Phet lab and worksheet.

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Stoichiometry Lab Report Chem 121L Part I: Introduction Stoichiometry is the study of the quantitative, or measurable, relationships that exist in chemical formulas and also chemical reactions. In this experiment hydrogen gas will be produced from the reaction of a known mass of magnesium metal with an excess of hydrochloric acid.

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Modified version of "Basic Stoichiometry Lab" by C. Bires. Useful for physical science introduction to limiting reactants. Subject Chemistry: Level High School, Middle School: Type Lab: Duration 30 minutes: Answers Included No: Language English: Keywords Limiting Reactants, Physical Science

[Physical Science Limiting Reactants Intro - PhET Contribution](#)

Reactants, Products and Leftovers

[Reactants, Products and Leftovers](#)

Lab HW Remote Discuss MC: Chemistry: Reactants, Products and Leftovers Activity 1: Intro to Chemical Reactions and Limiting Reactants: Trish Loeblein: UG-Intro HS MS: Remote Lab HW: Chemistry: Concept questions for Physics using PhET (Inquiry Based) Trish Loeblein: HS UG-Intro: MC: Physics

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Worksheet for Basic Stoichiometry (ANSWER 386.3g of LiNO3) 4) Using the following equation: Fe2O3 + 3 H2 ----> 2 Fe + 3 H2O. Calculate how many grams of iron can be made from 16.5 grams of Fe2O3 by the following equation. Worksheet for Basic Stoichiometry.